SCIE1000, $\quad$ Solutions to Tutorial Week 12.
(1) Here is an example of a hand-written answer which would receive full marks on an exam. (Pay careful attention to the solution; we didn't hand-write it because we were too lazy to type the answer. Instead, it should give you a good idea of how little you need to write, how 'messy' you can be, and how you can use abbreviations.)

201 exam Queshon 1.

1. One dose $=10^{6}+7 \times 10^{5}=17 \times 10^{5}$ units of virus

10 doses $/$ id $=17 \times 10^{6}$ virus units.
then 1 litre is $33.8 \times 17 \times 10^{6}$ units $=574.6 \times 10^{6}$

$$
\text { so } \frac{574.6 \times 10^{6}}{6.02 \times 10^{23}} \frac{\text { units } / \mathrm{L}}{\text { units } / \mathrm{mol}}=95.4 \times 10^{-17} \mathrm{~mol} / \mathrm{L}
$$

2. Many possible answers.

* 1 litre contains 338 dosiesí) so approx 300.
* 3 doses / person, so 1 litre treas 100 people.
*50 - if $N$ people have been vaccinated then $\frac{N}{100}$ L of OPV hove been used.
* how many people vocunoted? Current global pupratan about $7 \times 10^{9}$. If all vacanoted, $7 \times 10^{7} \mathrm{~L}$. If $1 / 3$ vacunated, bot $2 \times 10^{7} \mathrm{~L}$.

3. A. If 5 secondary infections ane, then a vacanotion rate $>80 \%$ on average means $<$ one of the 5 people will be susceptible and get sick, so the disease will due out.
B. $80 \% / 0.95=84.2 \%$.
C. For: wail save more lives and reduce suffering; gives a "safety margin". Against: expensive a difficult to do. Reduces money to
(2) The next two pages contain hand-written asnwers. Again, don't only pay attention to the correct answers; also think about how much you will write in your answers, and how you will set out your working.

2014 exam. Question 3.

1. In Year O (1988), 35200 cares, and this is constant $P_{0}$ in model $P_{0} e^{-k t}$

* Note. 1993 $\rightarrow 1997$, number approx halved. also $1997 \rightarrow 2001$ so half fe about 4 years.
Rule of 72 sos decay rate is approx $\frac{72}{4}=18 \%$
This is about -0.2 (decay).
So model is $35200 e^{-0.2 t}$

2. $t=-13$ (note negative). so $P=473923$.

Likely cause of large difference: is stated, UNICEF commenced eradication compoign to before the campotgnstarted.
3. $t=25$, so $P=237$. Difference probably a me to foot that $P(t)$ is just a model. We wald expect the actual figure to vary soreuhot from the modelled number, simply due to random (or real world) variation.

2014 exon Question 3 condinved.
4. We vat $P(l)=1$, so $35200 e^{-02 t}=1$
so $e^{-0.2 t}=\frac{1}{35200}$ so $-0.2 t=\ln \left(\frac{1}{35200}\right)$

so $t=52.35$, so year $(988+52.35$
= year 2040 (apron).
5.A. The function pred gives the modelled number of potto cases, calculating $p(t)$ fo year.
B. outfit: 2462.36

$$
111.27
$$

51.0

Total: 2624.62
C. For 3 years, finds the sum of the differences between the actual and predicted number of polio caves for each year, and in botel
D. "For each year, between actual and prechated polio cases".
(3) This is a discussion question.
(4) 1. For both hydrogen and iodine, the rate at which they are consumed in the reaction is dependent th the interaction between them (that is, it is proportional to the product of the two concentrations). The constant $-k$ in the first two equations is negative because the amounts of both hydrogen and iodine decrease as the reaction proceeds, and the constant $-k$ has the same value in both equations because the reaction requires one molecule of hydrogen for each molecule of iodine.
For hydrogen iodide, the increase in concentration is also dependent on the interaction between hydrogen and iodine, but is double the rate at which hydrogen and iodine individually are consumed. This is to be expected, as each reacting hydrogen molecule and iodine molecule form two hydrogen iodide molecules. The constant in this equation is positive, because the quantity of hydrogen iodide increases as the reaction proceeds.
2. To ensure conservation of mass and matter, the number of atoms in the solution (and hence the overall concentration) must remain constant. Thus, all $H_{2}$ and $I_{2}$ molecules consumed in the reaction must be converted to $H I$ molecules.
3. Consider the DE for the concentration of hydrogen:

$$
\left[H_{2}\right]^{\prime}=-k\left[H_{2}\right]\left[I_{2}\right]
$$

$\left[H_{2}\right]^{\prime}$ has units of $\mathrm{mol} \mathrm{L}^{-1} \mathrm{~s}^{-1}$, whilst both $\left[H_{2}\right]$ and $\left[I_{2}\right]$ have units of $\mathrm{mol}^{-1}$. Thus, to ensure that this DE is dimensionally balanced, $k$ must have units of $\mathrm{L} \mathrm{mol}^{-1} \mathrm{~s}^{-1}$.

1. At equilibrium there is no change in the concentration of reactants and products, so the values of $\left[H_{2}\right]^{\prime},\left[I_{2}\right]^{\prime}$ and $[H I]^{\prime}$ are all equal to zero.
2. Consider the DE for the rate of change of $H I$ at equilibrium:

$$
\begin{aligned}
{[H I]^{\prime} } & =2 k_{f}\left[H_{2}\right]\left[I_{2}\right]-2 k_{r}[H I]^{2} \\
\Longrightarrow 0 & =2 k_{f}\left[H_{2}\right]\left[I_{2}\right]-2 k_{r}[H I]^{2} \\
\Longrightarrow 2 k_{f}\left[H_{2}\right]\left[I_{2}\right] & =2 k_{r}[H I]^{2} \\
\Longrightarrow k_{f}\left[H_{2}\right]\left[I_{2}\right] & =k_{r}[H I]^{2}
\end{aligned}
$$

Remember that $2 k_{f}[H][I]$ represents the rate of formation of hydrogen iodide in the forward reaction. Hence $-2 k_{r}[H I]^{2}$ represents the rate of decomposition of hydrogen iodide in the reverse reaction. Then at equilibrium the rate of formation of $H I$ must equal the rate of decomposition of HI. (Note that it is possible to rearrange any of the DEs of the system at equilibrium to derive this equation.)
(6) 1. Substituting $z=2, x=1$ and $y=1$ into the expression for the equilibrium constant gives

$$
K=\frac{[H I]^{2}}{\left[H_{2}\right]\left[I_{2}\right]}
$$

Rearranging the equation in Part (e) gives

$$
\begin{aligned}
k_{f}\left[H_{2}\right]\left[I_{2}\right] & =k_{r}[H I]^{2} \\
\Longrightarrow \frac{k_{f}}{k_{r}} & =\frac{[H I]^{2}}{\left[H_{2}\right]\left[I_{2}\right]} \\
\Longrightarrow \frac{k_{f}}{k_{r}} & =K \\
\Longrightarrow K & =\frac{6.3 \times 10^{-2}}{1.8 \times 10^{-3}} \\
& =35
\end{aligned}
$$

2. We know that at a particular temperature, the value of the equilibrium constant is unaffected by initial concentrations. Hence

$$
K=35=\frac{[H I]^{2}}{\left[H_{2}\right]\left[I_{2}\right]}
$$

The hint tells us that $\left[H_{2}\right]=\left[I_{2}\right]$, so

$$
35=\frac{[H I]^{2}}{\left[H_{2}\right]^{2}}
$$

From the hint we know that the number of molecules in the system remains constant, and using the initial concentrations of all reactants/products gives:

$$
\begin{aligned}
{\left[H_{2}\right]+\left[I_{2}\right]+[H I] } & =\left[H_{2}\right]_{0}+\left[I_{2}\right]_{0}+[H I]_{0} \\
& =1+1+0 \\
& =2 \mathrm{~mol} \mathrm{~L}^{-1}
\end{aligned}
$$

Thus $\left[H_{2}\right]+\left[I_{2}\right]+[H I]=2$ so $[H I]=2-\left[H_{2}\right]-\left[I_{2}\right]$.
Remembering that the concentrations of $H_{2}$ and $I_{2}$ are equal at all times we can rewrite this expression for $[H I]$ as:

$$
[H I]=2-2\left[H_{2}\right]
$$

Substituting back into the expression for the equilibrium constant gives:

$$
\begin{aligned}
35 & =\frac{\left(2-2\left[H_{2}\right]\right)^{2}}{\left[H_{2}\right]^{2}} \\
35\left[H_{2}\right]^{2} & =4-8\left[H_{2}\right]+4\left[H_{2}\right]^{2} \\
31\left[H_{2}\right]^{2}+8\left[H_{2}\right]-4 & =0
\end{aligned}
$$

The hint tells us that $\left[H_{2}\right]=0.2527$ or -0.5107 , which must be $0.2527 \mathrm{~mol} \mathrm{~L}^{-1}$ because concentrations cannot be negative. Hence $\left[I_{2}\right]=0.2527 \mathrm{~mol} \mathrm{~L}^{-1}$ and $[H I]=2-2 \times 0.2527=1.4946 \mathrm{~mol} \mathrm{~L}^{-1}$.
(7) Here is an example of a hand-written answer which would receive full marks on an exam.
(a)(i) constant doubling tine

(ii) I at a $\downarrow$ rates power function
$\rightarrow$ eqn (S), with power $b<1$.

(8)

1. The expression $\frac{K-R}{K}$ models competition between rabbits, because it models a carrying capacity for the population. As the rabbit population increases, the this term approaches 0 , which means their growth rate, $R^{\prime}$, decreases.
2. The rate of change of the numbers of rabbits and foxes should be in units of rabbits/year and foxes/year. To be able to add and subtract the terms in the equations for $R^{\prime}$ and $F^{\prime}$, each term in the equation must have the same units. Thus, we have

$$
\operatorname{units}\left(R^{\prime}\right)=\operatorname{units}\left(a R \frac{K-R}{K}\right)=\operatorname{units}(b R F)
$$

and

$$
\operatorname{units}\left(F^{\prime}\right)=\operatorname{units}(c F)=\operatorname{units}(d R F)
$$

Looking at each of these equations individually gives,

$$
\begin{aligned}
& \frac{\text { rabbits }}{\text { year }}=\text { units }(a) \text { rabbits } \frac{\text { rabbits }- \text { rabbits }}{\text { rabbits }} \\
& =\text { units }(a) \text { rabbits } \\
& \text { year }{ }^{-1}=\text { units }(a) \\
& \frac{\text { rabbits }}{\text { year }}=\text { units }(b) \text { rabbits foxes } \\
& \text { year }{ }^{-1}=\text { units }(b) \text { foxes } \\
& (\text { year foxes) })^{-1}=\operatorname{units}(b) \\
& \frac{\text { foxes }}{\text { year }}=\text { units }(c) \text { foxes } \\
& \text { year }^{-1}=\operatorname{units}(c) \\
& \frac{\text { foxes }}{\text { year }}=\text { units }(d) \text { rabbits foxes } \\
& \text { year }{ }^{-1}=\text { units }(d) \text { rabbits } \\
& (\text { year rabbits })^{-1}=\operatorname{units}(d) .
\end{aligned}
$$

So the units of $a$ and $c$ are year ${ }^{-1}, b$ are (year foxes) ${ }^{-1}$, and $d$ are (year rabbits) ${ }^{-1}$.
3. If all the rabbits died suddenly, $R$ would become 0 . This means that the rate of change of foxes would be modelled by $F^{\prime}=-c F$. The rate of change of foxes is negative and proportional to the number of foxes, which means that the population of foxes is exponentially decaying. This makes sense, as the foxes would starve to extinction if their main food source becomes extinct. Here is a sketch of this behaviour:

4. If all the foxes died suddenly, $F$ would become 0 . This means that the rate of change of rabbits would be modelled by $R^{\prime}=-a R\left(\frac{K-R}{K}\right)$. This is the logistic model, so we know the population of rabbits will increase towards the carrying capacity $K$. This makes sense, as without their main predators, the rabbits will be free to live and breed to the point where the environment can just sustain the population. Here is a sketch of this behaviour:

5. Using these values,

$$
R^{\prime}=a R\left(\frac{K-R}{K}\right)-b R F=1 \times 100 \times\left(\frac{1000-100}{1000}\right)-0.1 \times 9 \times 100=90-90=0
$$

and

$$
F^{\prime}=-c F+d R F=-.1 \times 9+0.001 \times 9 \times 100=0.9-0.9=0 .
$$

Even without performing Euler's method, we can see that the populations won't change, since the rates of change of both the fox and rabbit populations are 0 . Performing one step of Euler's method anyway gives

$$
R=100+2 \times 0=100 \quad \text { and } \quad F=9+2 \times 0=9 .
$$

6. We found the rates of change to be 0 in part (e), which means the populations of rabbits and foxes under these conditions will remain constant. This means that for the values described in that question, the model is in perfect equilibrium. There are exactly enough rabbits to feed the foxes, but not allow their population to grow, and there are exactly enough foxes to keep the population of rabbits from increasing, but not eat so many that their population drops.
7. A. Using these values, we can find an expression for the rate of change of the fox population in terms of the rabbit population,

$$
F^{\prime}=-0.1 \times 20+0.001 \times 20 \times R=0.02 R-2 .
$$

So if $R=\frac{2}{0.02}=100$, then $F^{\prime}=0$ (i.e. the fox population remains constant). Furthermore, if $R>100$ then $F^{\prime}>0$ (i.e. the fox population increases), and if $R<100$ then the $F^{\prime}<0$ (i.e. the fox population decreases).
B. Again, using these values, we can find an expression for the rate of change of the rabbit population as a function of the current rabbit population,

$$
R^{\prime}=1 \times R \times\left(\frac{1000-R}{1000}\right)-0.1 \times 20 \times R=R(1-0.001 R-2)=-R(1+0.001 R) .
$$

Since the rabbit population is always positive, the term $R(1+0.001 R)$ is always positive, so $R^{\prime}$ will be negative, regardless of the actual value of $R$.
(9) Here is an example of a hand-written answer which would receive full marks on an exam.
(a)

(b)

$$
\begin{aligned}
& E^{\prime}=1000 \mathrm{~A}-E \\
& L^{\prime}=0.05 E-L \\
& P^{\prime}=0.1 L-P \\
& A^{\prime}=0.6 P-A
\end{aligned}
$$

note: 1000 2 becorve of one mole and ore femole $\rightarrow$ avge L000 eggs
eoch.
(c)

$$
\begin{aligned}
\text { at } t=0, E^{\prime} & =1000 \times 2-100 \\
& =1900 . \\
L^{\prime} & =0.05 \times 100-50=-45 \\
P^{\prime} & =0.1 \times 50-10=-5 \\
A^{\prime} & =0.6 \times 10-2=4
\end{aligned}
$$

Hence of $t=1$,

$$
\begin{aligned}
& \times 10-2=4 \\
& E=100+(900 \times 1=2000 . \\
& L=50-4 \times 1=5 \\
& P=10-5 \times 1=5 \\
& A=2+4 \times 1=6
\end{aligned}
$$

